

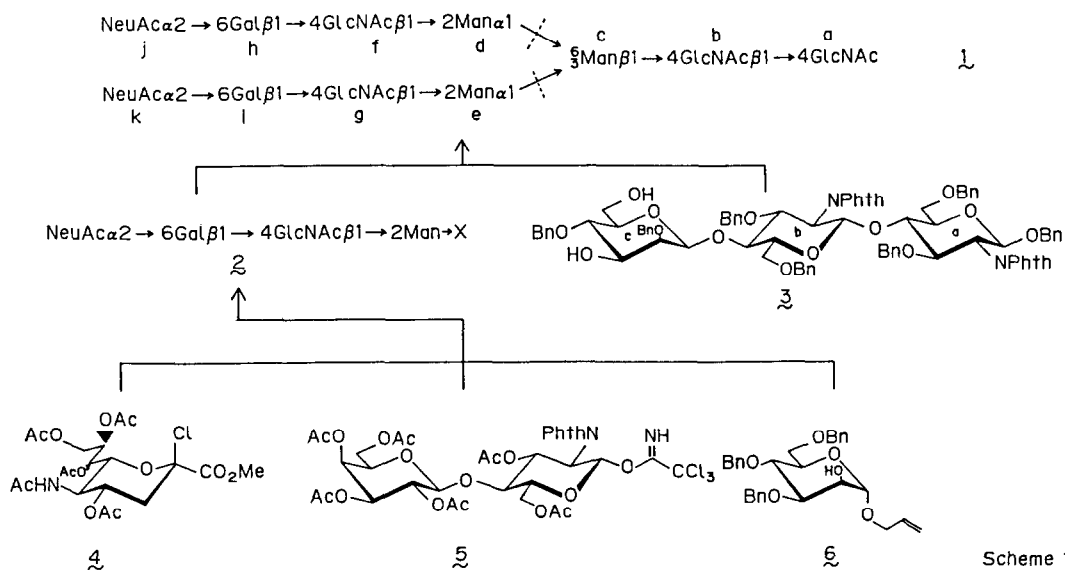
TOTAL SYNTHESIS OF A UNDECASACCHARIDE: A TYPICAL CARBOHYDRATE
 SEQUENCE FOR THE COMPLEX TYPE OF GLYCAN CHAINS OF A GLYCOPROTEIN¹⁾

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Abstract: A first total synthesis of undecasaccharide **1** of complex type of glycans of a glycoprotein was achieved in a stereo- and regiocontrolled way.

The glycan part of glycoproteins is now known to play significant roles for manifesting biological functions of the glycoproteins which control cellular regulation and recognition². The glycan **1** is classified as a typical complex type of glycans³ which is linked covalently to an Asn residue of proteins. As part of our project on the synthesis of glycans of glycoproteins, we describe here a total synthesis of the glycan **1**. In close connection with our project, an independent approach to the synthesis of N-linked glycans of glycoproteins has actively been pursued by Paulsen and his co-workers⁴.

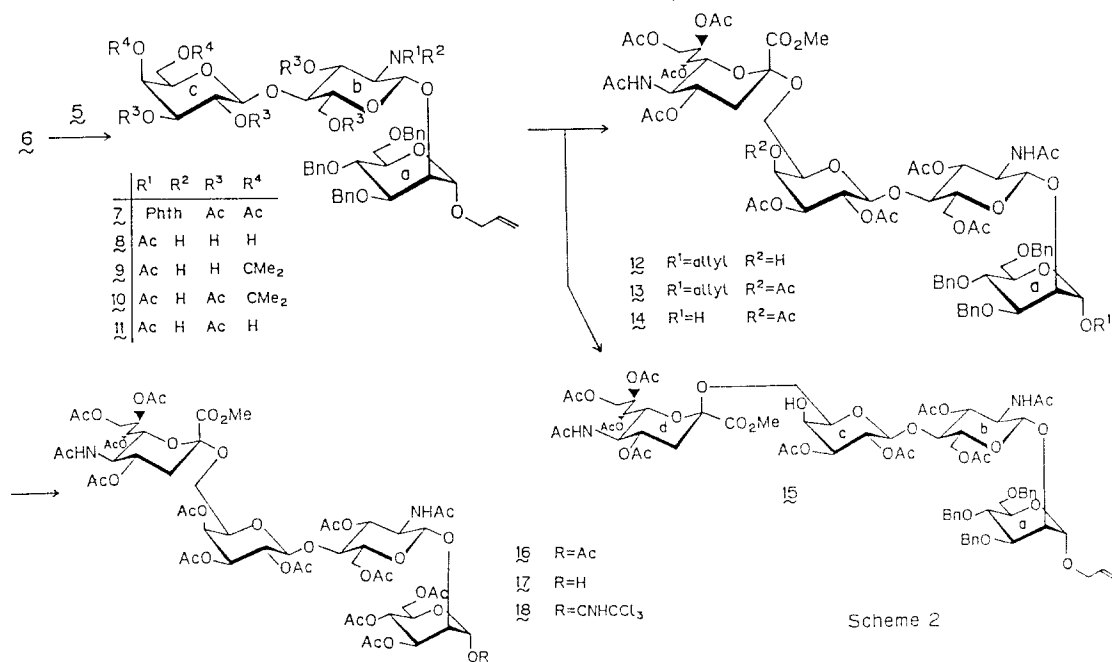
Retrosynthetic analysis of undecasaccharide **1** led us to design a glycotetraosyl donor **2** which correspond to two symmetric branches in **1**, and a glycosyl acceptor **3** which was prepared previously⁵. The glycotetraosyl donor **2** was planned to be synthesized from two glycosyl donors **4**⁶ and **5**⁷, and a glycosyl acceptor **6**⁸.



Scheme 1

Glycosylation of compound **6** with the imidate **5** in the presence of BF₃·Et₂O⁹-molecular sieves AW-300 in dichloroethane afforded a 73% yield of protected trisaccharide **7**, [α]_D -2.7° (c 1.3)¹⁰, Rf 0.38 in 2:1 toluene-EtOAc, δ_H ¹¹ 5.507 (d, 1 H, J 8.6 Hz, H-1b). Conversion of compound **7** into N-acetyl derivative **8**, Rf 0.41 in 3:1 CHCl₃-MeOH, δ_H 1.93 (s, 3 H, NAc), was

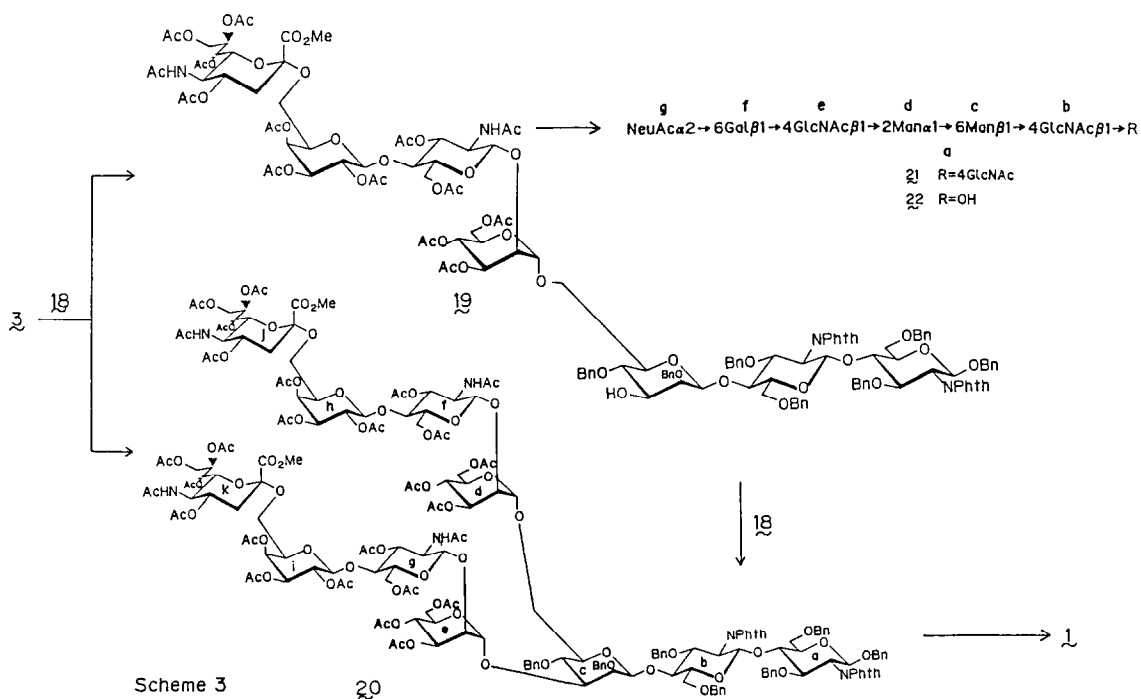
performed in 4 steps in 79% overall yield: (1) NaOMe-MeOH, (2) 1:5 nBuNH₂-MeOH,¹² reflux, (3) Ac₂O-pyridine, and (4) NaOMe-MeOH. Isopropylideneation of compound **8** with Me₂C(OMe)₂-TsOH-DMF at 0° followed by acetylation afforded a 67% yield of the desired kinetic product **10**, [α]_D +20.8° (c 0.9), R_f 0.35 in 1:3 toluene-EtOAc, δ_{H} 1.42 and 1.36 (2 s, 6 H, CMe₂), δ_{C} 98.94 (CMe₂), as well as a 14% yield of 3,4-O-isopropylidene isomer, R_f 0.44 in 1:3 toluene-EtOAc. Solvolysis of compound **10** in 1:1 AcOH-MeOH at 80° gave an 89% yield of diol **11**, [α]_D +11.3° (c 0.6), R_f 0.49 in 1:1 CH₂Cl₂-acetone, which was glycosylated with the chloride **4** in the presence of HgBr₂-Hg(CN)₂-molecular sieves 4A in Cl(CH₂)₂Cl. α -Linked product **12**, [α]_D -2.6° (c 0.9), R_f 0.44 in 10:1 CHCl₃-MeOH, δ_{H} 2.591 (dd, 1 H, J 4.7 and 12.9 Hz, H-3deq) was isolated in 48% yield along with β -isomer **15** (33%), [α]_D +2.1° (c 1.0), R_f 0.47 in 10:1 CHCl₃-MeOH, δ_{H} 2.464 (dd, 1 H, J 5.1 and 13.2 Hz, H-3deq).



Tetrasaccharide derivative **12** was transformed into a glycosyl donor **18**, a synthetic equivalent with the glycotetraosyl donor **2** depicted in Scheme 1. Acetylation of compound **12** gave a 95% yield of nonaacetate **13**, [α]_D -11.4° (c 0.9), which was deallylated with PdCl₂-NaOAc-aq.AcOH¹³ to give a 55% yield of hemiacetal **14**, [α]_D -19.8° (c 0.6), R_f 0.55 in 1:1 CH₂Cl₂-acetone. Hydrogenolysis of compound **14** in the presence of 10% Pd-C in 10:1 MeOH-AcOH, followed by acetylation afforded an 83% yield of peracetate **16**, [α]_D -25.5° (c 0.9), R_f 0.52 in 9:1 CHCl₃-MeOH, δ_{H} 6.019 (d, 1 H, J 1.5 Hz, H-1a), 4.866 (ddd, 1 H, J 4.6, 9.1, and 10.6 Hz, H-4d) and 2.537 (dd, 1 H, J 4.6 and 12.9 Hz, H-3deq). Chemoselective deacetylation of compound **16** with NH₂NH₂·AcOH in DMF¹⁴ and subsequent treatment of the resulted hemiacetal **17** with Cl₃CCN and DBU in Cl(CH₂)₂Cl afforded an 85% yield of the designed glycosyl donor **18**, [α]_D -19.4° (c 1.6), R_f 0.54 in 9:1 CHCl₃-MeOH, δ_{H} 8.694 (s, 1 H, =NH), and 6.203 (d, 1 H, J 1.7 Hz, H-1a), δ_{C} 95.1 (¹J_{CH} 175 Hz, C-1a).

Having the key glycosyl donor **18** prepared, we now describe crucial glycosylation of the key acceptor **3** with the donor **18**. We expected that glycosylation at a primary hydroxyl group

(C-6c OH) of compound **3** may proceed smoothly but the reaction at C-3c OH might be sluggish. Therefore, selective monoglycosylation of the acceptor **3** was first examined. A mixture of compound **3** and 0.77 equivalent of the glycosyl donor **18** was stirred in the presence of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ -molecular sieves AW-300 at -10° and flash chromatography of the products over silica gel in 25:1 CHCl_3 -MeOH afforded a 48% yield (based on the donor **18**) of the monoglycosylated product **19**, $[\alpha]_D -12.2^\circ$ (c 1.0), Rf 0.62 in 9:1 CHCl_3 -MeOH. 400 MHz ^1H -nmr revealed signals of 43 protons for aromatic protons and 42 protons for 14 acetyl methyl protons as well as a singlet for methyl ester at δ 3.794, supporting that the product **19** was a monoglycosylated one. The structure of compound **19** was further confirmed by conversion into free heptasaccharide **21**, $[\alpha]_D +7.7^\circ$ (c 0.1, H_2O), Rf 0.18 in 2:1:1 nBuOH-EtOH- H_2O in 4 steps in 44% overall yield: (1) LiI-pyridine, 120° , (2) $\text{NH}_2\text{NH}_2 \cdot \text{H}_2\text{O}$ -EtOH, reflux, (3) Ac_2O -MeOH, and (4) 10% Pd-C in 7:3 MeOH- H_2O . 400 MHz ^1H -Nmr of compound **21** in D_2O showed following signals: δ (60°) 5.175 (d, 0.5 H, J 2.4 Hz, H-1 α), 4.906 (s, 1 H, H-1d), 4.726 (s, 1 H, H-1c), 4.685 (d, 0.5 H, J 8.0 Hz, H-1 α), 4.595 (m, 2 H, H-1b and H-1e), 2.663 (dd, 1 H, J 4.4 and 12.2 Hz, H-3geq), 2.057, 2.040, 2.033, 2.019 (4 s, 12 H, 4NAc), and 1.667 (t, 1 H, J 12.5 Hz, H-3gax), in good agreement with the data for related natural hexasaccharide **22**¹⁵.



Since regio- and stereochemistry of compound **19** was established, further elongation of a glycan chain at C-3c OH of the heptasaccharide **19** was examined and, under the same condition by using 1.3 equivalents of the glycosyl donor **18**, was obtained, after purification by high pressure gel permeation chromatography (GPL 220 column, HITACHI) in CHCl_3 , a 56% yield of the desired protected undecasaccharide **20**, $[\alpha]_D -7.8^\circ$ (c 0.4), Rf 0.12 in 1:1 CCl_4 -acetone, δ_{H} 3.803 and 3.789 (2 s, 6 H, OMe). Compound **20** was deblocked in 4 steps as in the case of **19** to give free undecasaccharide **1**, $[\alpha]_D -4.6^\circ$ (c 0.05, H_2O), Rf 0.14 in 2:1:1 nBuOH-EtOH- H_2O in 15% overall yield. 400 MHz ^1H -Nmr of compound **1** in D_2O contained signals at δ (60°) 5.185 (d, 0.5

H, J 2.2 Hz, H-1a α), 5.128 (s, 1 H, H-1e), 4.925 (s, 1 H, H-1d), 4.753 (s, 1 H, H-1c), 4.695 (d, 0.5 H, J 7.4 Hz, H-1a β), 4.602 (d, 3 H, J 7.5 Hz, H-1b, H-1h and H-1g), 4.226 (H-2c), 4.173 (H-2e), 4.090 (H-2d), 2.674 (dd, 2 H, J 4.2 and 12.2 Hz, H-3jeq. and H-3keq), and 1.679 (t, 2 H, J 11.7 Hz, H-3jax and H-3kax). These observed data for synthetic **1** was found to be in good agreement with related oligosaccharides isolated from glycoproteins¹⁵.

In conclusion, a regio- and stereocontrolled total synthesis of undecasaccharide **1**, a typical carbohydrate sequence for complex type of glycans of glycoproteins, was achieved for the first time. Comparison of ¹H-nmr of synthetic **1** and **21** with those of naturally occurring glycans isolated from glycoproteins provided unambiguous synthetic evidences for the proposed structures of this class of glycans.

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REFERENCES AND NOTES

- 1) Part 51 in the series "Synthetic Studies on Cell Surface Glycans". For part 50, see M. Sugimoto, M. Numata, K. Koike, Y. Nakahara, and T. Ogawa, submitted for Carbohydr. Res.
- 2) N. Sharon and H. Lis, in H. Neurath and R. L. Hill (Eds.), "The proteins" Vol. 5, Academic Press, New York, pp 1-144 (1982).
- 3) R. Kornfeld and S. Kornfeld, Annu. Rev. Biochem., **45**, 217 (1976); J. Montreuil, Pure and Appl. Chem., **42**, 431 (1975); Adv. Carbohydr. Chem. Biochem., **37**, 158 (1980); E. G. Berger, E. Buddecke, J. P. Kamerling, A. Kobata, J. C. Paulson, and J. F. G. Vliegenthart, Experientia, **38**, 1129 (1982).
- 4) H. Paulsen, Chem. Soc. Rev., **13**, 15 (1984).
- 5) T. Ogawa, T. Kitajima, and T. Nukada, Carbohydr. Res., **123**, C5 (1983).
- 6) R. Kuhn, P. Lutz, and D. L. MacDonald, Chem. Ber., **99**, 611 (1966).
- 7) R. R. Schmidt and G. Grundler, Angew. Chem. Int. Ed. Engl., **22**, 776 (1983).
- 8) T. Ogawa, K. Beppu, and S. Nakabayashi, Carbohydr. Res., **93**, C6 (1981); T. Ogawa and T. Nukada, ibid., **136**, 135 (1985).
- 9) R. R. Schmidt and J. Michel, Angew. Chem. Int. Ed. Engl., **19**, 731 (1980); R. R. Schmidt, J. Michel, and M. Roos, Liebigs Ann. Chem., 1343 (1984).
- 10) Values of $[\alpha]_D$ were measured for CHCl₃ solutions at 25°, unless noted otherwise. Compounds with $[\alpha]_D$ recorded gave satisfactory data for elemental analyses.
- 11) Values of δ_C and δ_H were expressed in p.p.m. downwards from the signal for internal Me₄Si, for solutions in CDCl₃, unless noted otherwise.
- 12) P. L. Durette, E. P. Meitzner, and T. Y. Shen, Tetrahedron Lett., 4013 (1979); Carbohydr. Res., **77**, C1 (1979).
- 13) R. Bose and R. Scheffold, Angew. Chem., **88**, 578 (1976); T. Ogawa and S. Nakabayashi, Carbohydr. Res., **93**, C1 (1981).
- 14) G. Excoffier, D. Gagnaire, and T.-P. Utille, Carbohydr. Res., **39**, 368 (1975).
- 15) J. F. G. Vliegenthart, L. Dorland, and H. van Halbeek, Adv. Carbohydr. Chem. Biochem., **41**, 209 (1983).

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